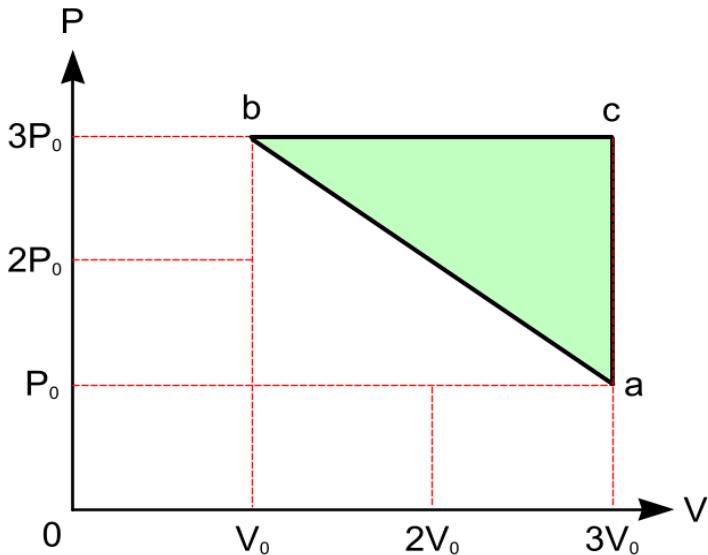


1. Engine. Use $PV = nRT$, $P_0V_0 = nRT_0$, and take the number of moles to be such that

$$nR = 1 \frac{\text{joules}}{\text{kelvins}} = 1 \frac{J}{K} . \text{ Take the temperature } T_0 = 300\text{K} .$$



- a. Show that $T_a = 900\text{K}$.
- b. Show that $T_b = 900\text{K}$.
- c. Show that $T_c = 2700\text{K}$.
- d. Show that $\Delta Q_{a \rightarrow b} = -1200 \text{ J}$
- e. Show that $\Delta Q_{b \rightarrow c} = 4500 \text{ J}$
- f. Show that $\Delta Q_{c \rightarrow a} = -2700 \text{ J}$.
- g. Calculate $\eta = \frac{W_{\text{net}}}{Q_{\text{in}}}$.

2. Stat Mechanics. The energy levels are $0, \varepsilon, 2\varepsilon, 3\varepsilon, 4\varepsilon, 5\varepsilon, \dots$, with $\varepsilon = kT$. Use the identity

$$S = 1 + r + r^2 + r^3 + r^4 \dots = \frac{1}{1-r}, \text{ where } r < 1, \text{ to show that the partition function}$$

$$Z = \sum_{n=0}^{\infty} e^{-\frac{E_n}{kT}}$$

is 1.58. Find the occupation probability in each of these levels: $0, \varepsilon, 2\varepsilon, 3\varepsilon, 4\varepsilon$.

3. Quantum Mechanics. The time independent Schrödinger equation in one dimension can be

$$\text{written as } H\psi = E\psi \text{ where } H = -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + V(x) . \text{ Let } V(x) = \frac{1}{2}m\omega^2x^2 . \text{ Then consider}$$

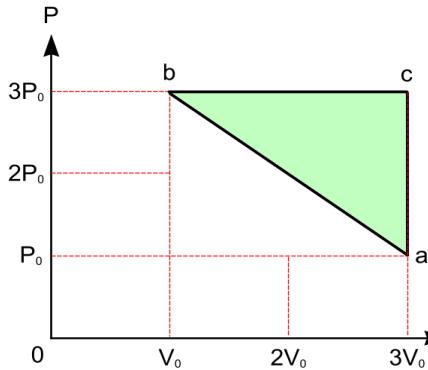
special units where $\hbar = m = \omega = 1$. Show that $\psi(x) = \frac{1}{\sqrt{\pi}}e^{-\frac{x^2}{2}}$ is an eigenfunction, i.e.,

$$H\psi = E\psi , \text{ and find the eigenvalue } E.$$

4. Eigenvectors and Eigenvalues. Find the normalized eigenvectors and eigenvalues for the

$$\text{matrix operator } A = \begin{bmatrix} 1 & 2 \\ 3 & 0 \end{bmatrix} . \text{ A vector } \begin{bmatrix} c_1 \\ c_2 \end{bmatrix} \text{ is normalized if } c_1^*c_1 + c_2^*c_2 = 1 .$$

1. Engine. Ideal gas with number of moles so that $PV = T$. Also, $P_0 V_0 = T_0 = 300\text{K}$.



- a. At a, $P_a V_a = T_a = P_0(3V_0) = 3T_0 = 900\text{K}$.
- b. At b, $P_b V_b = T_b = (3P_0)V_0 = 3T_0 = 900\text{K}$.
- c. At c, $P_c V_c = T_c = (3P_0)(3V_0) = 9T_0 = 2700\text{K}$.
- d. $\Delta U_{ab} = 0$, $\Delta Q_{ab} = \Delta W_{ab} = -4P_0 V_0 = -1200 \text{ J}$,
using Trapezoid Area: (average height $2P_0$)(base of $2V_0$).
- e. $\Delta U_{bc} = \frac{3}{2}nR\Delta T_{bc} = \frac{3}{2}\Delta T_{bc} = \frac{3}{2}6T_0 = 9 \cdot 300 = 2700 \text{ J}$

$$\Delta W_{bc} = 3P_0 2V_0 = 6P_0 V_0 = 6T_0 = 1800 \text{ J} \text{ and } \Delta Q_{bc} = \Delta U_{bc} + \Delta W_{bc} = 2700 + 1800 = 4500 \text{ J}$$

$$f. \Delta W_{ca} = 0 \text{ so } \Delta Q_{ca} = \Delta U_{ca} = \frac{3}{2}\Delta T_{ca} = \frac{3}{2}(T_a - T_c) = \frac{3}{2}(900 - 2700) = -2700 \text{ J}$$

$$g. \eta = \frac{W_{\text{net}}}{Q_{\text{in}}} = \frac{W_{\text{enclosed area}}}{Q_{\text{in}}} = \frac{(2P_0)(2V_0)/2}{Q_{bc}} = \frac{2T_0}{4500} = \frac{600}{4500} = \frac{6}{45} = \frac{2}{15}$$

2. Stat Mech. The energy levels are $0, \varepsilon, 2\varepsilon, 3\varepsilon, 4\varepsilon, 5\varepsilon, \dots$, with $\varepsilon = kT$. Let $r = e^{-1}$.

$$Z = \sum_{n=0}^{\infty} e^{-\frac{E_n}{kT}} = e^0 + e^{-1} + e^{-2} + e^{-3} \dots = 1 + r + r^2 + r^3 + \dots = \frac{1}{1-r} = \frac{1}{1-e^{-1}} = \frac{e}{e-1} = 1.58$$

$$\frac{n_0}{N} = \frac{e^0}{Z} = \frac{1}{1.58} = 0.63, \quad \frac{n_1}{N} = \frac{e^{-1}}{1.58} = 0.23, \quad \frac{n_2}{N} = \frac{e^{-2}}{1.58} = 0.09, \quad \frac{n_3}{N} = \frac{e^{-3}}{1.58} = 0.03, \quad \frac{n_4}{N} = \frac{e^{-4}}{1.58} = 0.01$$

3. Quantum Mech. $H = -\frac{1}{2} \frac{d^2}{dx^2} + \frac{1}{2}x^2$ and $\psi(x) = \frac{1}{\sqrt{\pi}} e^{-\frac{x^2}{2}}$. Show $H\psi = E\psi$. Find E .

$$-\frac{1}{2} \frac{d^2\psi}{dx^2} = \left[-\frac{1}{2} \frac{d^2}{dx^2} \right] \frac{1}{\sqrt{\pi}} e^{-\frac{x^2}{2}} = \frac{1}{2} \frac{d}{dx} \left[x \frac{1}{\sqrt{\pi}} e^{-\frac{x^2}{2}} \right] = \frac{1}{2} \left[\frac{1}{\sqrt{\pi}} e^{-\frac{x^2}{2}} \right] - \frac{1}{2} \left[\frac{x^2}{\sqrt{\pi}} e^{-\frac{x^2}{2}} \right]$$

$$-\frac{1}{2} \frac{d^2\psi}{dx^2} = \frac{1}{2}\psi - \frac{1}{2}x^2\psi, \quad -\frac{1}{2} \frac{d^2\psi}{dx^2} + \frac{1}{2}x^2\psi = \frac{1}{2}\psi. \text{ Eigenfunction } \psi, \text{ eigenvalue } \frac{1}{2}.$$

4. Eigenvectors and Eigenvalues. $A = \begin{bmatrix} 1 & 2 \\ 3 & 0 \end{bmatrix}$, $\det \begin{bmatrix} 1-\lambda & 2 \\ 3 & -\lambda \end{bmatrix} = 0$, $\lambda^2 - \lambda - 6 = 0$

$$(\lambda - 3)(\lambda + 2) = 0 \text{ gives eigenvalues 3 and -2.}$$

$$\begin{bmatrix} 1 & 2 \\ 3 & 0 \end{bmatrix} \begin{bmatrix} c_1 \\ c_2 \end{bmatrix} = 3 \begin{bmatrix} c_1 \\ c_2 \end{bmatrix} \text{ gives } c_1 + 2c_2 = 3c_1, \quad c_2 = c_1, \text{ normalized eigenvector } u = \frac{1}{\sqrt{2}} \begin{bmatrix} 1 \\ 1 \end{bmatrix}.$$

$$\begin{bmatrix} 1 & 2 \\ 3 & 0 \end{bmatrix} \begin{bmatrix} c_1 \\ c_2 \end{bmatrix} = -2 \begin{bmatrix} c_1 \\ c_2 \end{bmatrix} \text{ gives } c_1 + 2c_2 = -2c_1, \quad c_2 = -\frac{3}{2}c_1 \text{ leading to } v = \frac{1}{\sqrt{13}} \begin{bmatrix} 2 \\ -3 \end{bmatrix}.$$